# TECHNIC, INC. SAFETY DATA SHEET

# **SECTION 1. IDENTIFICATION**

Product Name: SILFLAKE<sup>®</sup> 600 Product Code: 95-600 Recommended use: Conductive filler; for industrial use only. Supplier: Technic Inc. 1 Spectacle Street Cranston, RI 02910 USA Telephone no: 401-781-6100 Emergency no: ChemTrec 800-424-9300, Outside the USA and Canada 703-527-3887

# **SECTION 2. HAZARD IDENTIFICATION**

#### **Classification:**

Flammable Liquids, Category 4 Eye Damage/Irritation, Category 2A Chronic Aquatic Toxicity, Category 1

#### Label elements and precautionary statements:

Signal word: Warning

Pictogram(s):



#### Hazards not otherwise classified: None

#### Hazard statement(s):

Combustible liquid Causes serious eye irritation Very toxic to aquatic life with long lasting effects

#### Precautionary statement(s):

Keep away from heat, sparks, open flames and hot surfaces. No smoking.

Wear protective gloves, clothing and eye and face protection.

In case of fire: Use foam, dry chemical or carbon dioxide (CO2) to extinguish.

If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

If eye irritation persists: Get medical advice/attention.

Avoid release to the environment.

Store in a well-ventilated place. Keep cool.

Collect spillage.

Dispose of contents and container in accordance with local, state and federal regulations.

# SECTION 3. COMPOSITION AND INFORMATION ON INGREDIENTS

Chemical name	CAS Number	EINECS Number	Concentration
Silver (metallic)	7440-22-4	231-131-3	84-90%
Naphtha solvent	64742-94-5	265-198-5	10-16%

# SECTION 4. FIRST AID MEASURES

#### Inhalation:

Remove patient to fresh air. Support breathing if required. Obtain medical treatment for dizziness, unconsciousness or irritation or difficulty in breathing.

#### Skin contact:

Remove contaminated clothing and wash affected area thoroughly with soap and water. Launder clothing before rewearing. Seek medical attention for prolonged skin irritation.

#### Eye contact:

Flush with water, including under lids, for fifteen minutes. Obtain immediate medical attention.

#### Ingestion:

If patient is conscious, rinse particles from mouth with water; do not swallow. DO NOT induce vomiting. Never give anything by mouth to an unconscious person. Obtain immediate medical attention.

## **SECTION 5. FIRE-FIGHTING MEASURES**

**Suitable extinguishing media:** Use only powder approved for Class D (metal) fires. Cool endangered containers with water spray.

Unsuitable extinguishing media: Do not use water on metal (Class D) fires.

**Specific hazards in case of fire:** Grinding finely divided powder, particularly with strong oxidizers, may result in an explosion or fire when mixed with air in the proper proportions. Toxic metal fumes may be released in a fire situation.

**Special protective equipment and precaution for fire fighters:** For fires in enclosed areas, wear self-contained breathing apparatus and full protective gear. Do not inhale combustion gases.

# SECTION 6. ACCIDENTAL RELEASE MEASURES

#### Personal precautions, protective equipment and emergency procedures:

Wear appropriate skin, eye and respiratory protection. Avoid contact with eyes. Avoid prolonged or repeated skin contact and breathing dust or powder. Use in a well-ventilated area. Do not eat, drink or smoke while cleaning up. Ensure adequate ventilation.

#### Methods and materials for containment and cleaning up:

Wear appropriate personal protective gear. Collect spilled material for silver recovery by vacuuming or sweeping without raising dust. Rinse spill area with water. Do not allow this material or its rinsings to enter storm or sanitary sewers or other waterways (see also Section 13).

#### Environmental precautions:

Prevent spills and rinsings from entering storm or sanitary sewers or other waterways and contact with soil.

## SECTION 7. HANDLING AND STORAGE

#### Precautions for safe handling:

Avoid contact with eyes. Avoid prolonged repeated skin contact and breathing mists or vapors. Use in well-ventilated area. Do not empty waste into sanitary drains.

#### Conditions for safe storage, including incompatibilities:

Store in a cool, dry area. Use with adequate ventilation. Keep container tightly closed when not in use. Store only in the original container. Empty containers may contain residual flammable solvent vapors.

## SECTION 8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Exposure Limits:

<u>Ingredient</u> Silver (metallic) ACGIH TLV 0.1 mg/m<sup>3</sup> TWA OSHA PEL 0.01 mg/m<sup>3</sup> Other Limits 0.01 mg/m<sup>3</sup> NIOSH REL

#### Appropriate engineering controls:

Use in well-ventilated area with local exhaust.

#### **Respiratory protection:**

Wear appropriate, approved respirator when ventilation is inadequate to meet exposure limits.

**Eye protection:** Chemical splash goggles or safety glasses with side shields must be worn.

#### Skin protection:

Wear rubber or neoprene gloves. Wear rubber apron and long sleeves to prevent skin contact. Wash hands thoroughly with soap and water after handling and before eating or smoking.

## **SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES**

Physical state: Metal powder in solvent Color: White, lustrous metallic powder; Clear, colorless solvent **Odor:** Aromatic Odor threshold: Not available pH: Not applicable Melting/freezing point: 960 °C (1760 °F) metal; -18 °C (0 °F) solvent Initial boiling point: Not determined Flash point: 66 °C (151 °F) solvent Evaporation rate: Not applicable Flammability (solid, gas): Not applicable **Upper/lower explosion limits:** 5.9%/0.8% (solvent) Vapor pressure: Not determined Vapor density: Not determined **Relative density (H<sub>2</sub>O = 1) @25 °C:** 10.5 (metal); 0.896 (solvent) Solubility: Insoluble in water and other common solvents Partition coefficient octanol/water: Not applicable Auto-ignition temperature: 445 °C (833 °F) solvent Decomposition temperature: Not determined Viscosity: Not applicable

## SECTION 10. STABILITY AND REACTIVITY

**Reactivity:** Stable, non-reactive when stored and used according to recommendations.

Chemical stability: No decomposition if used according to specifications.

Possibility of hazardous reactions: None are known.

Conditions to avoid: Open flames and other sources of ignition.

Incompatible materials: Strong acids, bases and oxidizing agents; ammonia, acetylene and hydrogen peroxide.

Hazardous decomposition products: Oxides of carbon and hydrocarbon fragments due to incomplete combustion.

## SECTION 11. TOXICOLOGICAL INFORMATION

#### **Routes of Exposure and Symptoms**

Inhalation: May cause respiratory tract irritation.

- Ingestion: May cause non-specific symptoms of gastric upset due to presence of particulate material.
- **Skin Contact:** Short-term contact produces no symptoms. Prolonged and repeated contact may cause argyria (see below).

**Eye Contact:** May cause eye irritation which is usually reversible.

Acute and Chronic Effects from Short- and Long-term Exposure: See Routes of Exposure and Symptoms above.

Acute Oral Toxicity:	TDLo:	330 mg/kg (rat, silver metal)
	LD50:	>5000 mg/kg (rat, naphtha solvent)

- Acute Dermal Toxicity: LD50: >2000 mg/kg (rabbit, naphtha solvent
- Acute Inhalation Toxicity: LC50: 4688 mg/m<sup>3</sup> (rat, naphtha solvent)

Acute Eye Irritation: Particulate material may cause temporary irritation. Naphtha solvent may cause mild, short-lasting discomfort.

**Dermal Irritation:** Prolonged exposure to high concentrations can lead to argyria, a generalized grayish pigmentation of the skin and mucous membranes. Such symptoms usually occur after at least two years of exposure. There are no systemic effects, other symptoms, or physical disabilities known to be associated with this condition.

#### Carcinogen Listings:

IARC: No NTP: No OSHA: No

Reproductive Effects: No applicable information available.

Target Organ Effects: No applicable information available.

## **SECTION 12. ECOLOGICAL INFORMATION**

Aquatic Toxicity: Silver is one of the most toxic metals known to aquatic organisms.

Persistence and degradability: Silver is not biodegradable and is toxic to aquatic life. Solvent is readily biodegradable.

**Bio-accumulative potential:** There is evidence to suggest bioaccumulation of silver will occur.

**Mobility in soil:** Accidental spillage may lead to penetration in the soil and groundwater. Improper handling and disposal of this material may cause environmental damage.

## SECTION 13. DISPOSAL CONSIDERATIONS

#### Waste Disposal:

Disposal of this material is subject to user compliance with applicable laws and regulations and consideration of product characteristics at time of disposal.

## **SECTION 14. TRANSPORT INFORMATION**

Not regulated by DOT (road, rail), IMDG (sea) or IATA/ICAO (air)

Marine Pollutant: No

## **SECTION 15. REGULATORY INFORMATION**

#### **Inventory Status:**

All components are on TSCA, EINECS/ELINCS, AICS, and DSL.

#### U.S. Regulations:

U.S. Superfund Amendments and Reauthorization Act (SARA) Title III:

#### SARA (311/312) HAZARD CATEGORIES:

\_\_\_\_None \_\_\_Immediate \_\_\_Delayed \_\_X\_Fire \_\_\_Reactive \_\_\_Pressure generating
SARA 313: This product contains the following SARA 313 Toxic Release Chemicals.
Chemical Name CAS Number Concentration

Silver (metallic)

**CAS Number** 7440-22-4

**Concentration** 84-90%

The following product components are cited on the lists below:Chemical NameCAS NumberNoneList CitationsCalifornia Proposition 65 List

## SECTION 16. OTHER INFORMATION

**VOC (Volatile Organic Compounds):** Naphtha solvent, 10-16%

HMIS Ratings:

Health: 1 Flammability: 0 Reactivity: 0 Personal Protection: C

Prepared by: Allan H. Reed

**SDS Preparation date:** May 12, 2016 **Supersedes previous version:** New SDS.

This SDS contains revisions in the following section(s): Not applicable.New SDS.

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End of Safety Data Sheet